

DTU



# Quantifying climate change impacts of CCS and CCU at waste incinerators

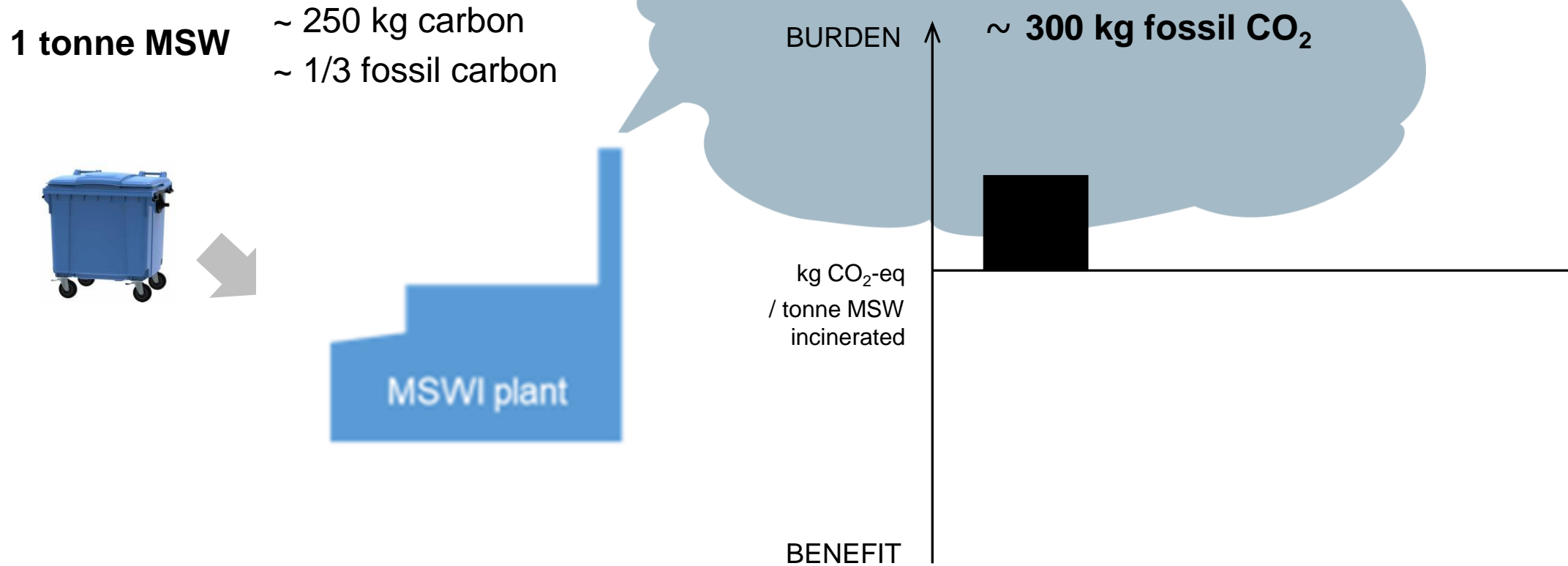
T.H. Christensen and V. Bisinella

DTU Sustain, Department of Environmental Engineering, Technical University of Denmark

*In collaboration with Christian Riber and Tore Hulgaard, Rambøll Consultants and Anders Damgaard, DTU*



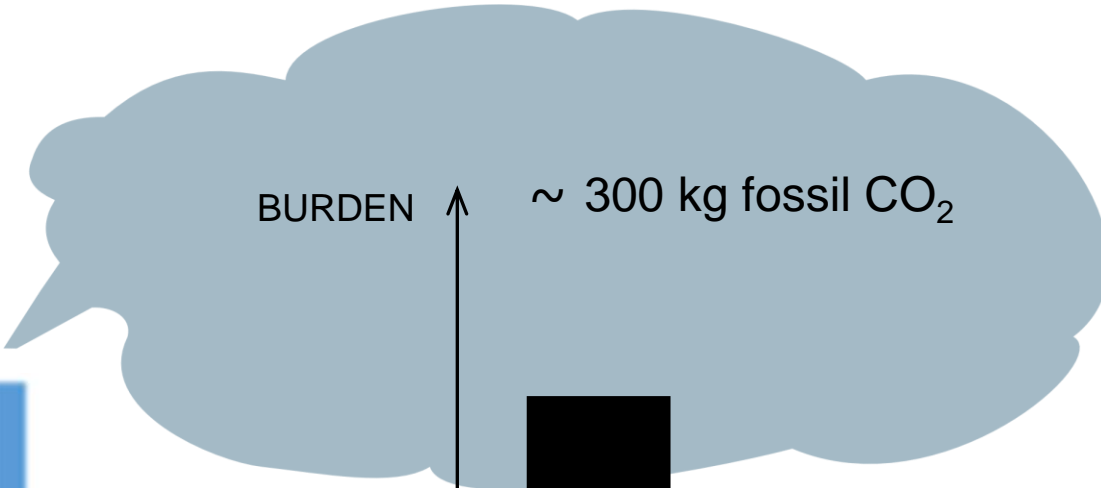
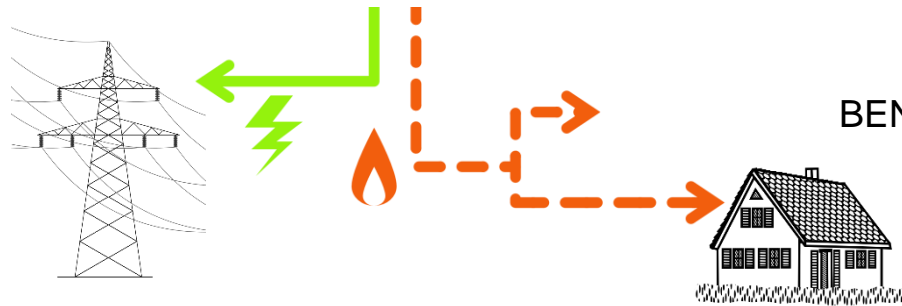
# Waste incineration is a contributor to climate change



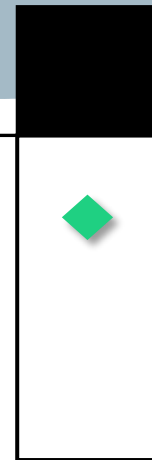
# Climate change impacts and energy recovery

**1 tonne MSW**

- ~ 250 kg carbon
- ~ 1/3 fossil carbon
- ~ **8-11 GJ LHV**



kg CO<sub>2</sub>-eq / tonne MSW incinerated



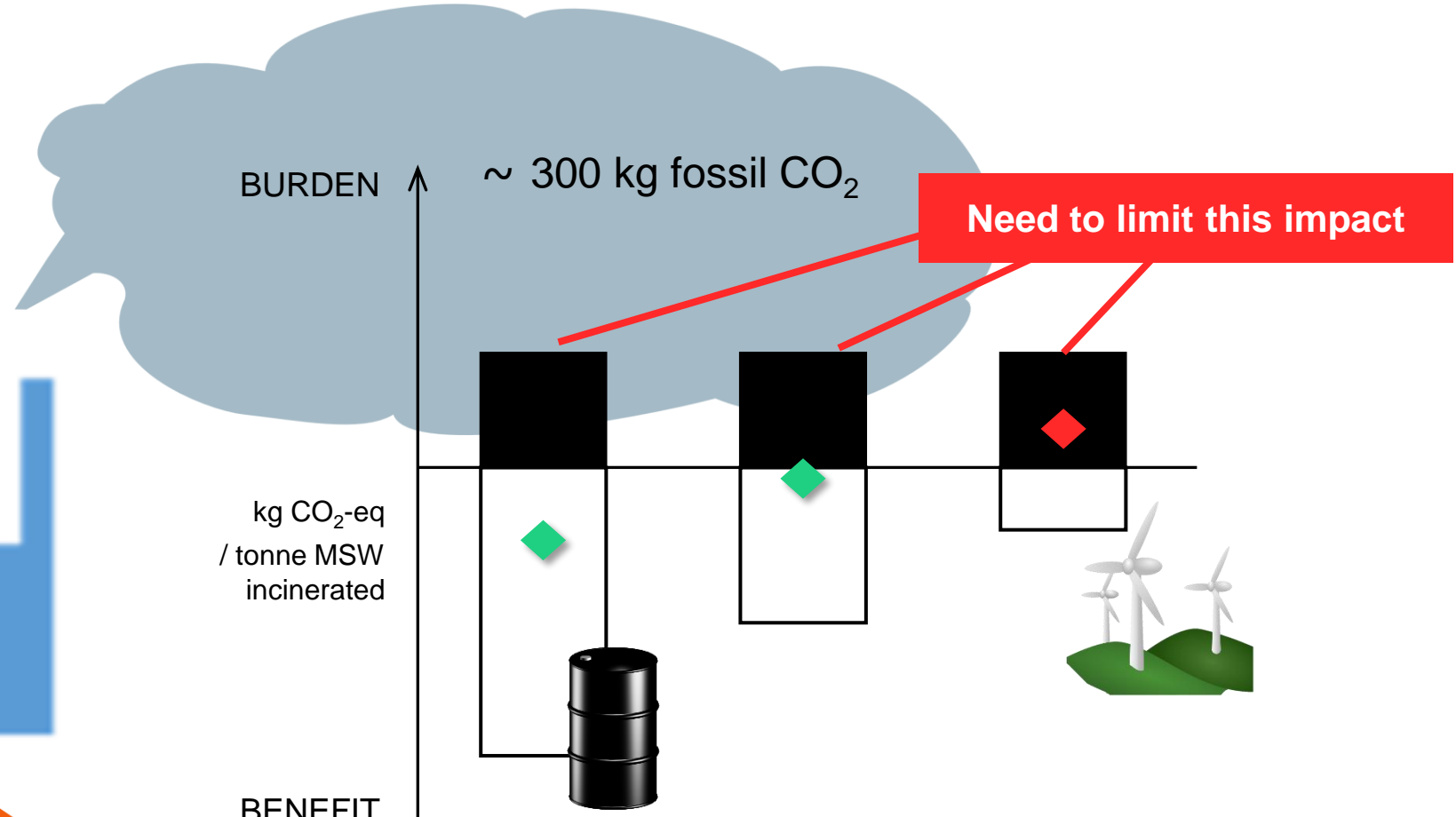
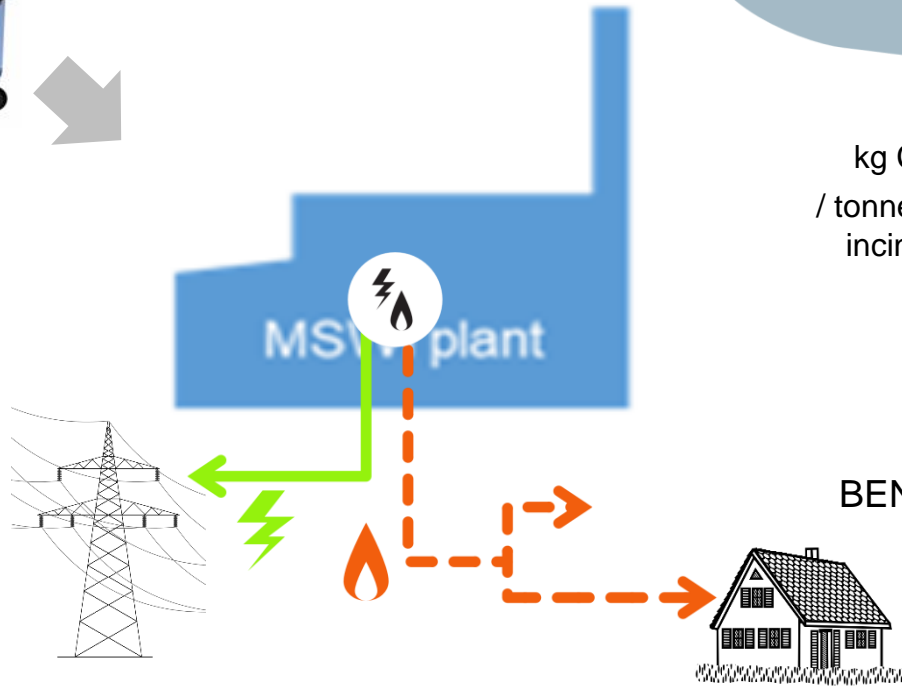
BENEFIT

~ 2-8 GJ/tonne MSW

**Depends on incinerator configuration and waste composition**

# Net climate change contribution in the future

**1 tonne MSW**  
 ~ 250 kg carbon  
 ~ 1/3 fossil carbon  
 ~ 8-11 GJ LHV

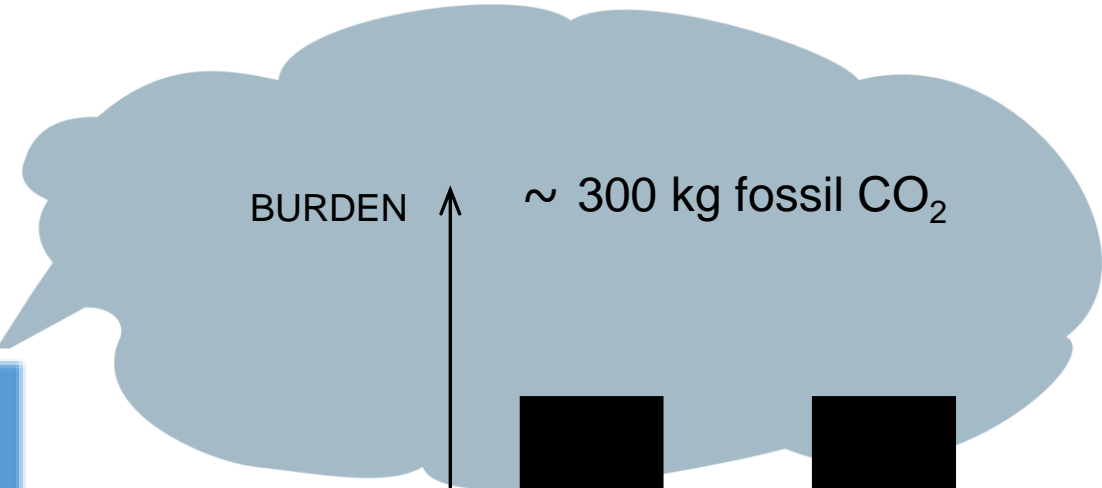
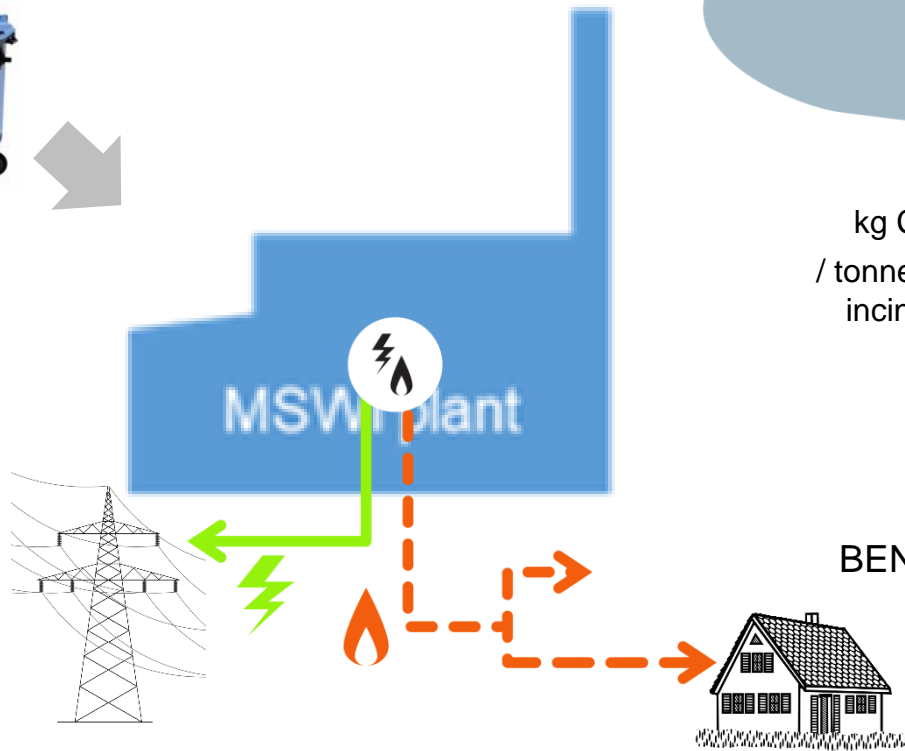


**Offset and climate change impact or benefit depend on energy sources substituted**  
 e.g. kg CO<sub>2</sub>-eq/kWh

# Net climate change contribution in the future

1 tonne MSW

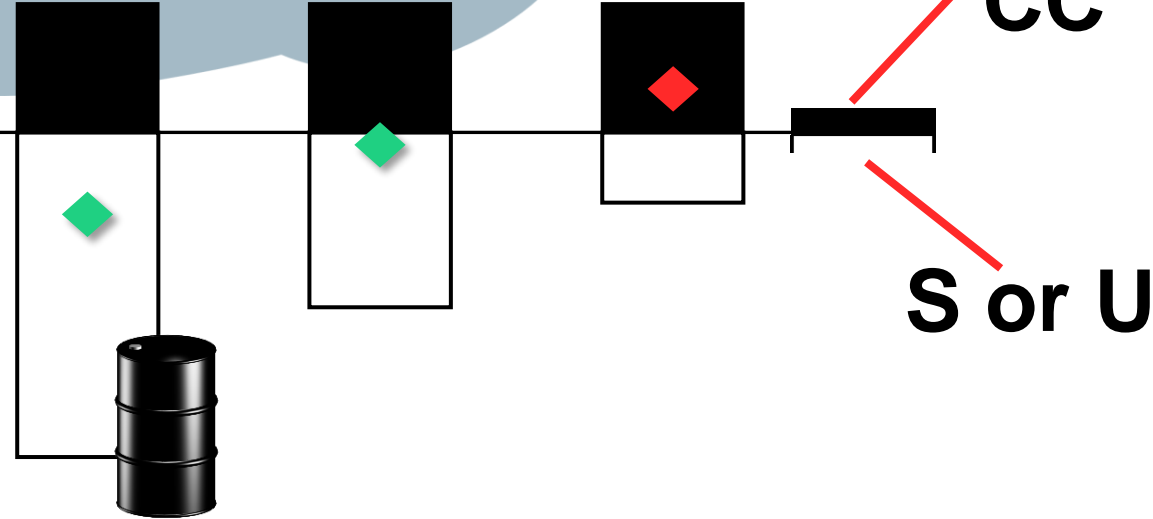
- ~ 250 kg carbon
- ~ 1/3 fossil carbon
- ~ 8-11 GJ LHV



BURDEN

kg CO<sub>2</sub>-eq / tonne MSW incinerated

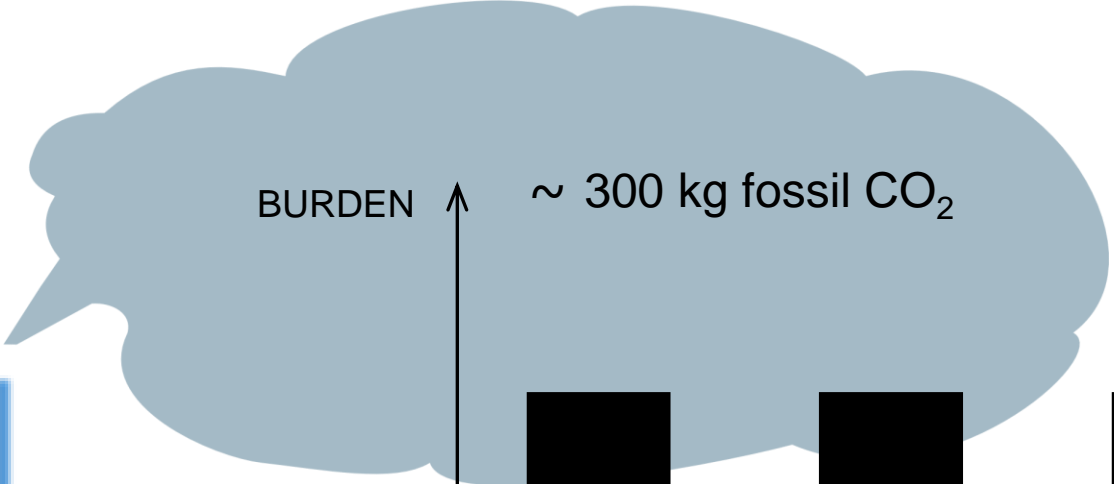
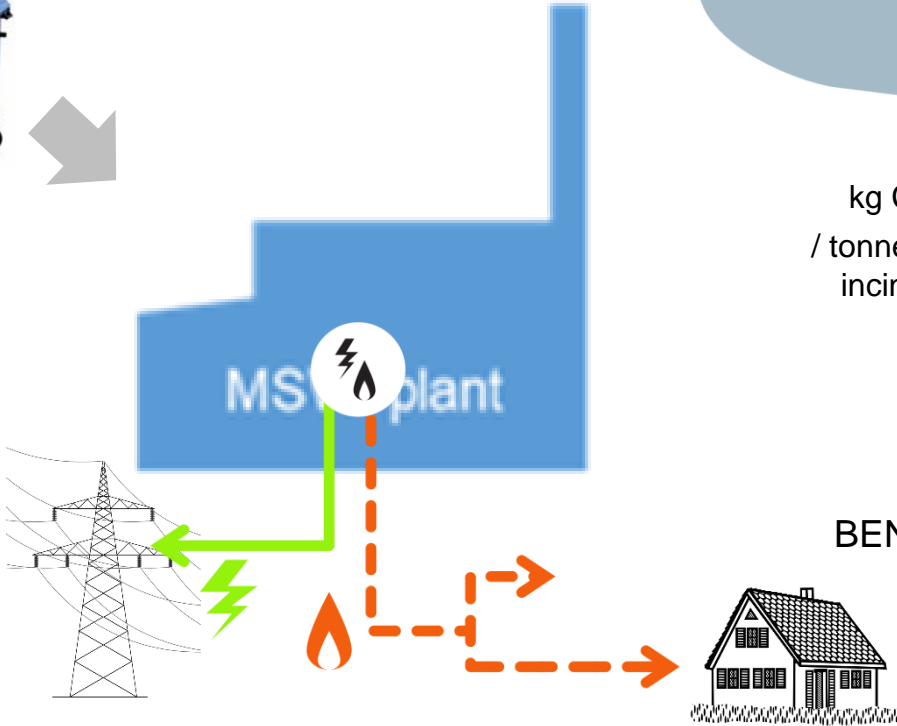
BENEFIT



# Net climate change contribution in the future

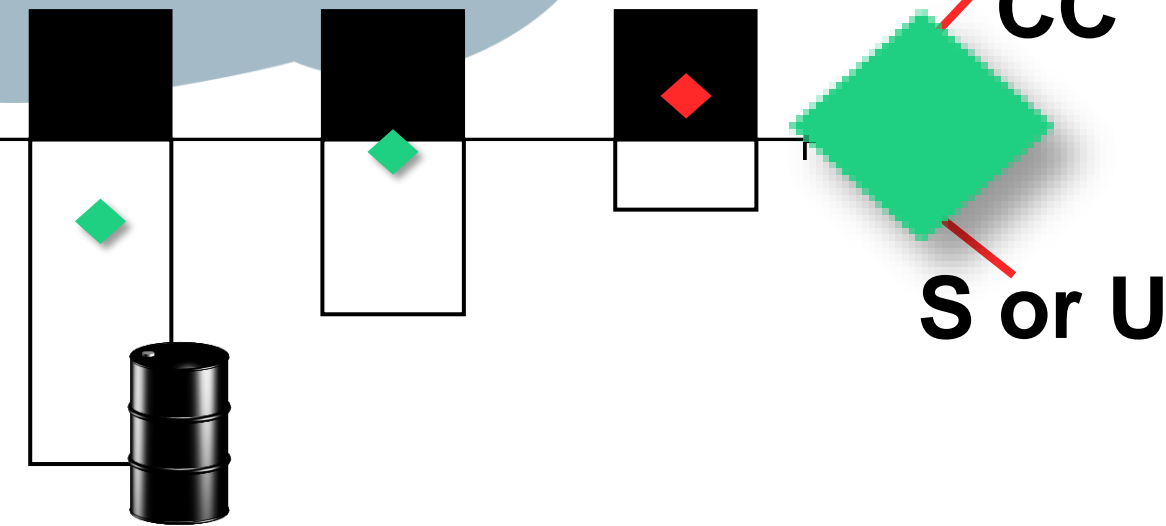
1 tonne MSW

- ~ 250 kg carbon
- ~ 1/3 fossil carbon
- ~ 8-11 GJ LHV



kg CO<sub>2</sub>-eq / tonne MSW incinerated

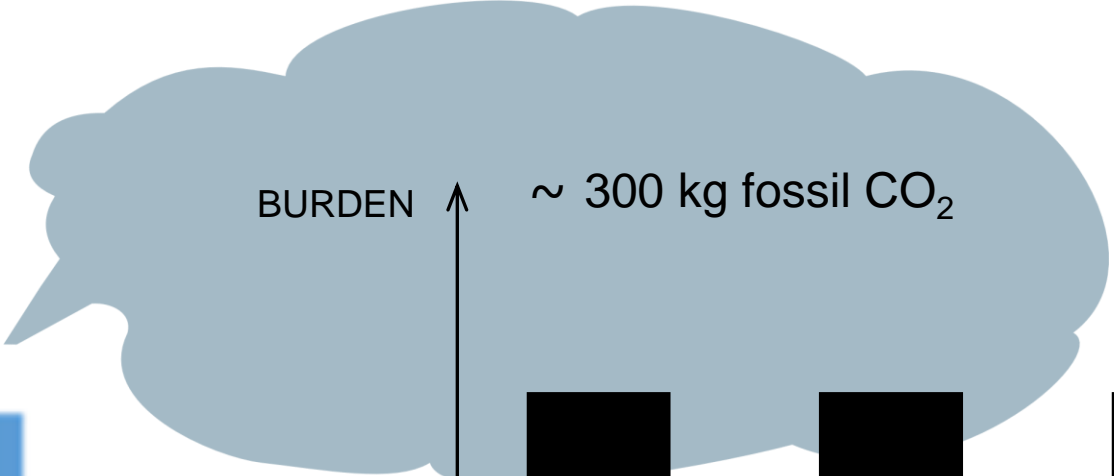
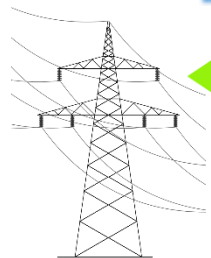
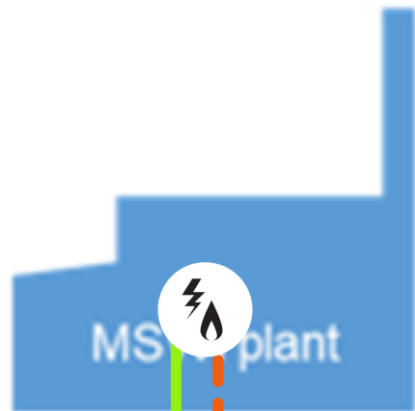
BENEFIT



# Net climate change contribution in the future

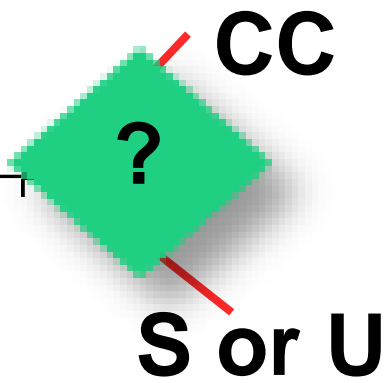
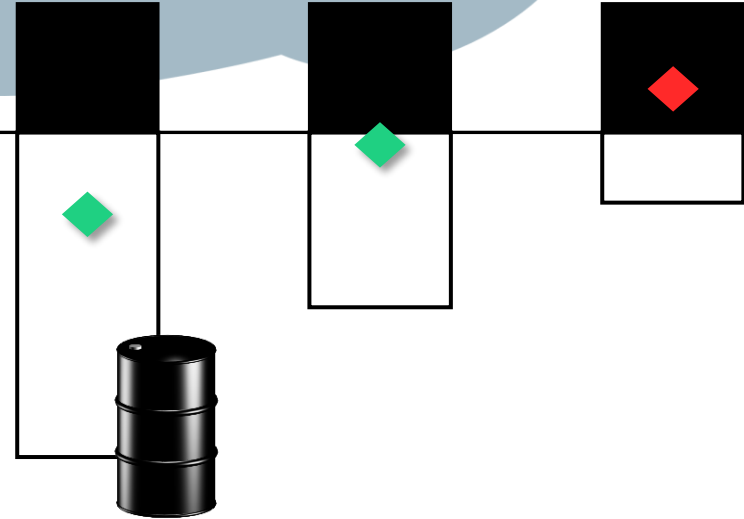
1 tonne MSW

- ~ 250 kg carbon
- ~ 1/3 fossil carbon
- ~ 8-11 GJ LHV



BURDEN ↑  
kg CO<sub>2</sub>-eq / tonne MSW incinerated

BENEFIT ↓







# Aim: Climate change impacts

## **Assess the impacts of amending MSWI with CC**

How can CC apply to MSWI? What are the implications in terms of energy recovery for different MSWI configurations?

## **Assess the impacts of management options for the captured CO<sub>2</sub>**

- Storage
- Utilization:

## **Effects of the exchange with the energy system being in transition**

How does the climate change impacts CCS and CCU change in different energy systems in which the technology operates?

# Methods

- Life Cycle Assessment carried out in EASETECH
- Foreground system modelling: new MSWI with/without CC processes (9 configurations)
- Background data from Ecoinvent 3.6 (ancillary materials and energy)
  
- Waste composition varied
- Energy flows modelled
  
- Analysis of results:
  - CO<sub>2</sub> balance
  - Energy recovery in terms of electricity and heat
  - Process contribution analysis
  - Scenario analysis (energy system, utilization options)
  - Sensitivity analysis (perturbation)

# Methodological assumptions: Carbon

**39% of the carbon in the waste is fossil**

**61% of the carbon in the waste is biogenic**

Climate change impacts are quantified according to the following characterization factors:

- 0 kg CO<sub>2</sub>-eq/kg CO<sub>2</sub> for biogenic CO<sub>2</sub> emissions (e.g. from food waste)
- 1 kg CO<sub>2</sub>-eq/kg CO<sub>2</sub> for fossil CO<sub>2</sub> emissions (e.g. non-recyclable plastic and synthetic textiles)
- -1 kg CO<sub>2</sub>-eq/kg CO<sub>2</sub> for avoided emission of biogenic CO<sub>2</sub>
- 0 kg CO<sub>2</sub>-eq/kg CO<sub>2</sub> for avoided emissions of fossil CO<sub>2</sub>

Similarly for chemicals and fuels substituted

We keep track of the fossil/biogenic content of emissions, products and fuels

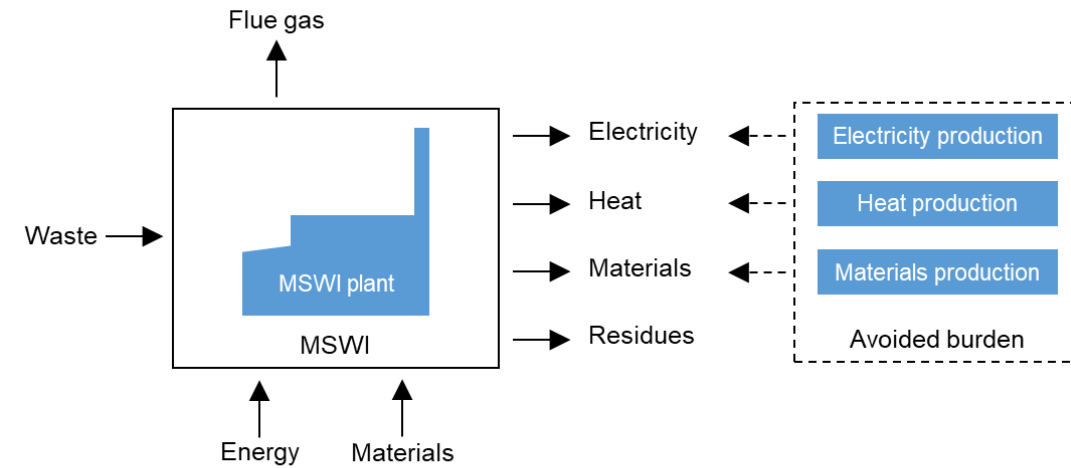
# MSWI configurations

MSW varied 9.3 - 11.3 GJ/Tonne

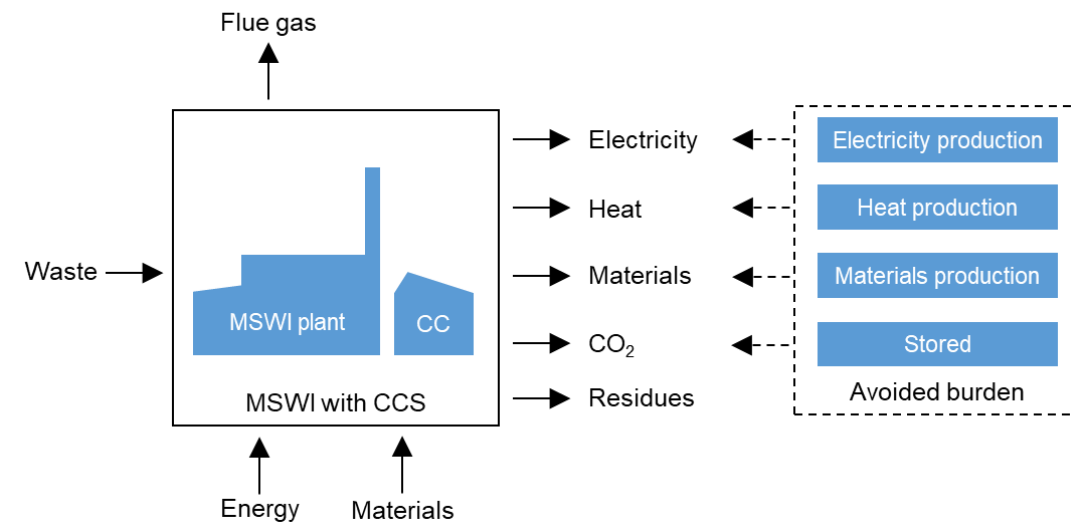
MSWI technology varied:

- Flue gas condensation
- Heat/ no heat recovery
- w/wo CC
- w/wo heat recovery from CC unit

MSWI without carbon capture (A - E)



MSWI with carbon capture (F - I)



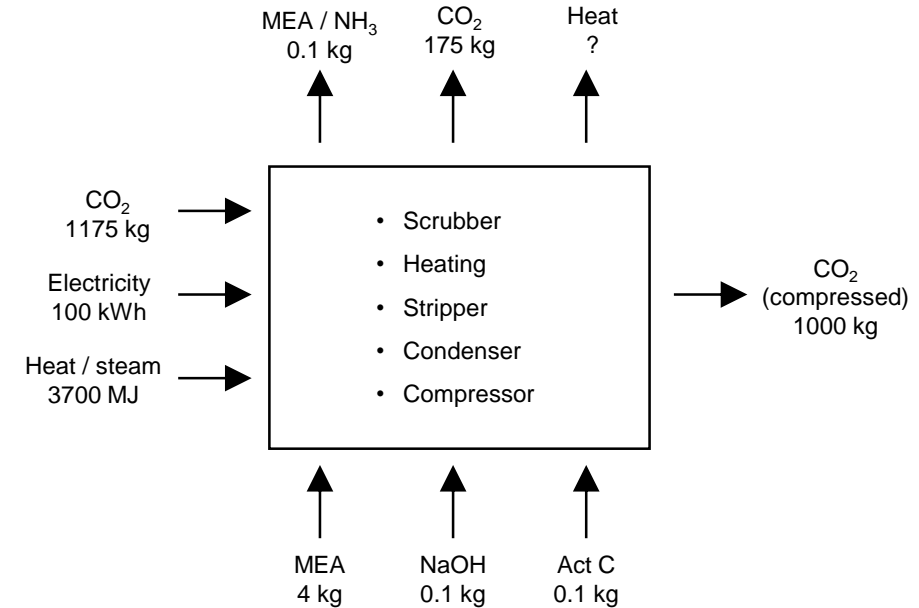
# Carbon capture (CC) technology

- **MEA:**

CO<sub>2</sub> is exposed to a monoethanolamine (MEA) solution forming a water soluble salt. After stripping at 100 - 140 °C and condensing of water, the >99% purity CO<sub>2</sub> is compressed. About 85-90% of the CO<sub>2</sub> is captured. The MEA is recovered and recycled, but some losses must be compensated. The electricity consumption is low, but the steam use for the MEA recovery is high (**3700 MJ per tonne CO<sub>2</sub>**) after Reiter and Lindorfer (2017). The technology is well established.

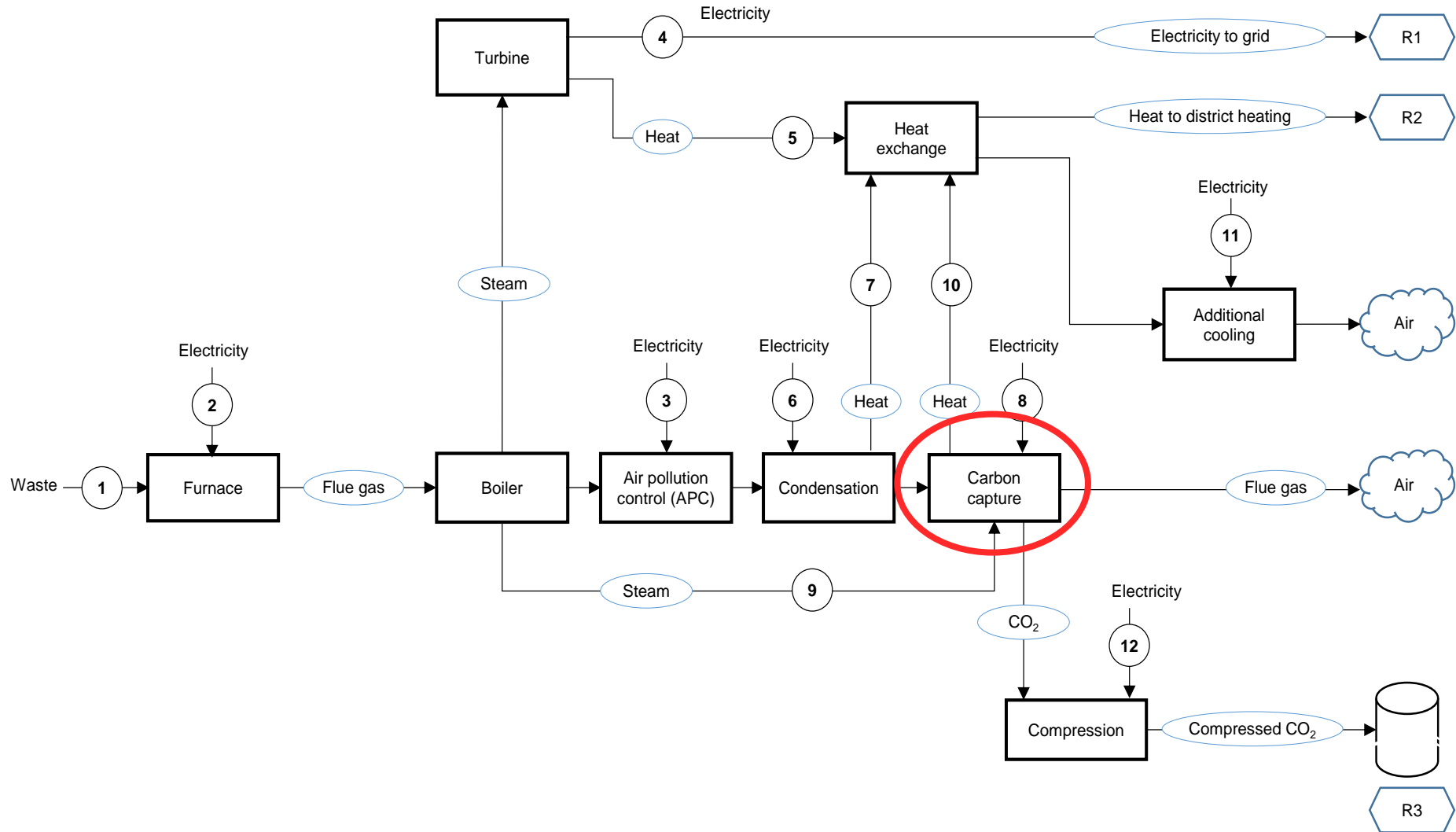
- **Other:**

- PSA (pressure-swing operation)
- Hydrate-based
- Cryogenic distillation
- Membrane filtration



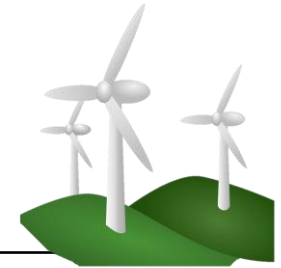


# CC amended to MSWI



# Energy system scenarios

## What if the energy system changes?



SCENARIO No.	1	2	3	4	5	6 BASELINE	7	8	9	10	11	12
Notes	Hypothetical	Hypothetical	Hypothetical	NETP 6DS single 2020-2030	NETP 6DS mix 2020-2030	NETP 4DS mix 2020-2030	Hypothetical	NETP 2DS mix 2020-2030	NETP 4DS single 2020-2030	Hypothetical	Hypothetical	Hypothetical
<b>ELECTRICITY</b>												
<b>Share</b>	%	%	%	%	%	%	%	%	%	%	%	%
Oil	100											
Hard coal		100			17							
Natural gas			100	100	51	23		17				
Biomass with LUC							100					
Solar panels						45		29	100			
Wind onshore					23	19		15		100		
Wind offshore					9	13		39			100	100
<b>kg CO<sub>2</sub>-eq/kWh</b>	<b>1.20</b>	<b>1.04</b>	<b>0.70</b>	<b>0.70</b>	<b>0.54</b>	<b>0.21</b>	<b>0.21</b>	<b>0.16</b>	<b>0.09</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>
<b>HEAT</b>												
<b>Share</b>	%	%	%	%	%	%	%	%	%	%	%	%
Hard coal	100											
Oil		100										
Natural gas			100									
Biomass with LUC				100	100	54	100	42	100	100	100	
Heat pumps						46		58				100
<b>kg CO<sub>2</sub>-eq/MJ</b>	<b>0.13</b>	<b>0.09</b>	<b>0.07</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.02</b>	<b>0.01</b>

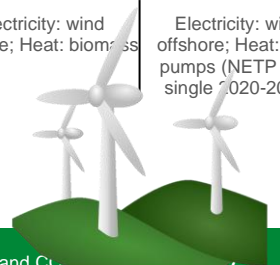
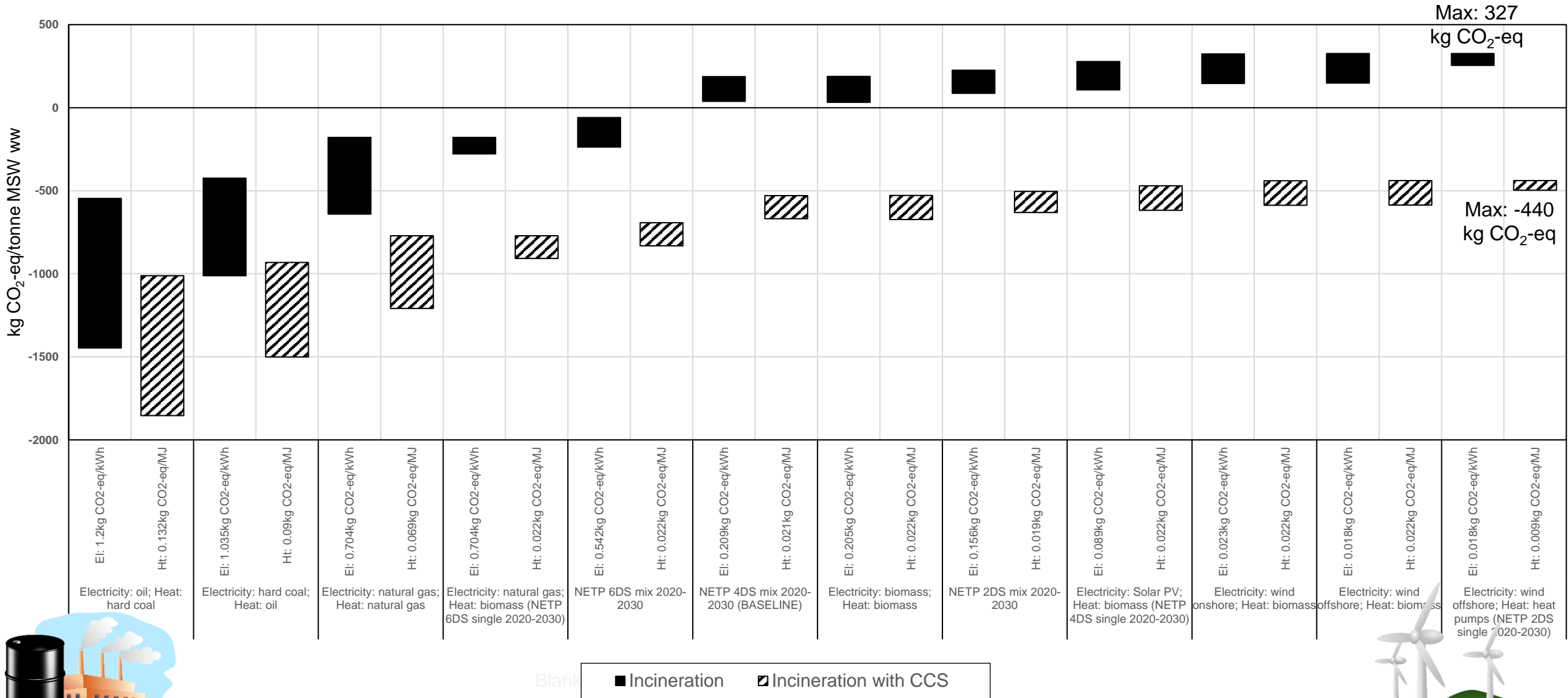
50-60 times reduction

10-13 times reduction

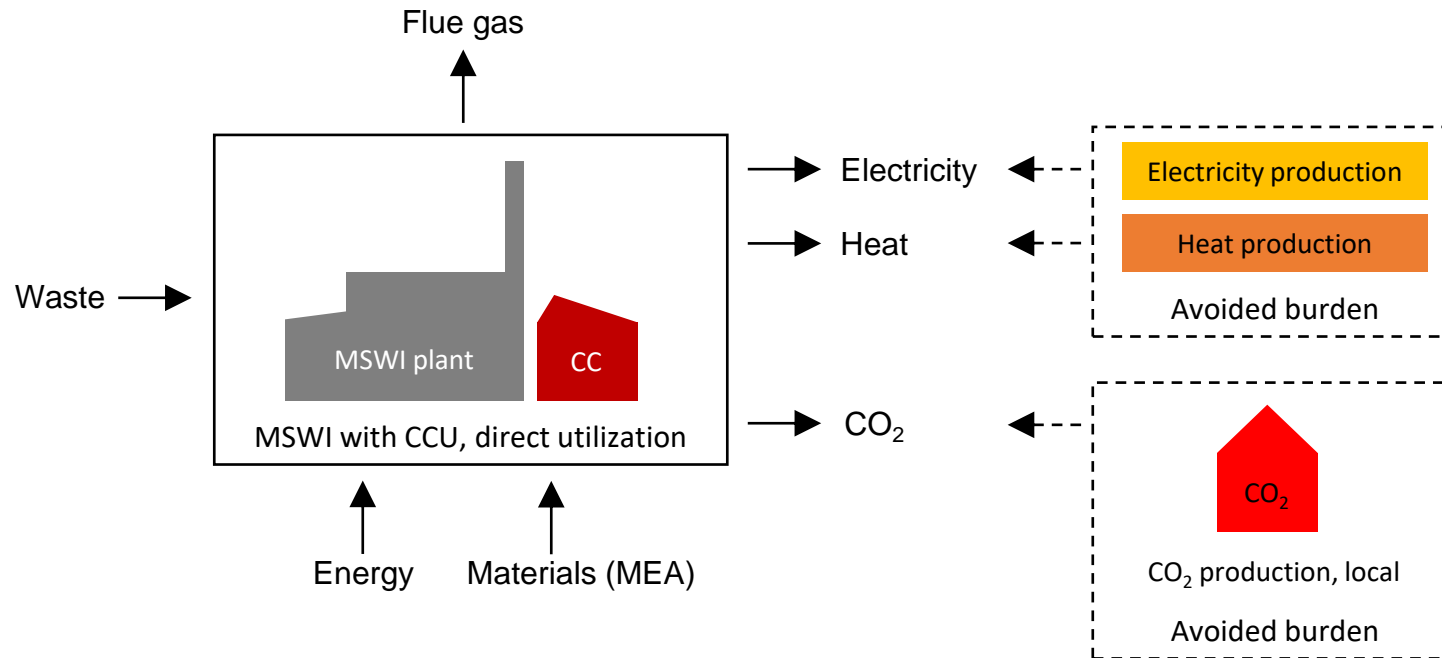
# Storage

- Captured CO<sub>2</sub> is transported and stored
- Transportation:
  - Different options (road tankers, railroad tankers, sea carriers, pipeline)
  - Environmental impacts: energy used for compression, emissions of CO<sub>2</sub> during loading/unloading and transport
  - Data from industries document losses of CO<sub>2</sub> 1-2% per 1000 km transported
  - We assumed 1000 km distance and 1% loss
- Storage:
  - Geological formations
  - We assume that CO<sub>2</sub> is stored indefinitely (> 1000 years)
  - Loss 2% CO<sub>2</sub>

# CCS: Climate change impacts: different energy systems

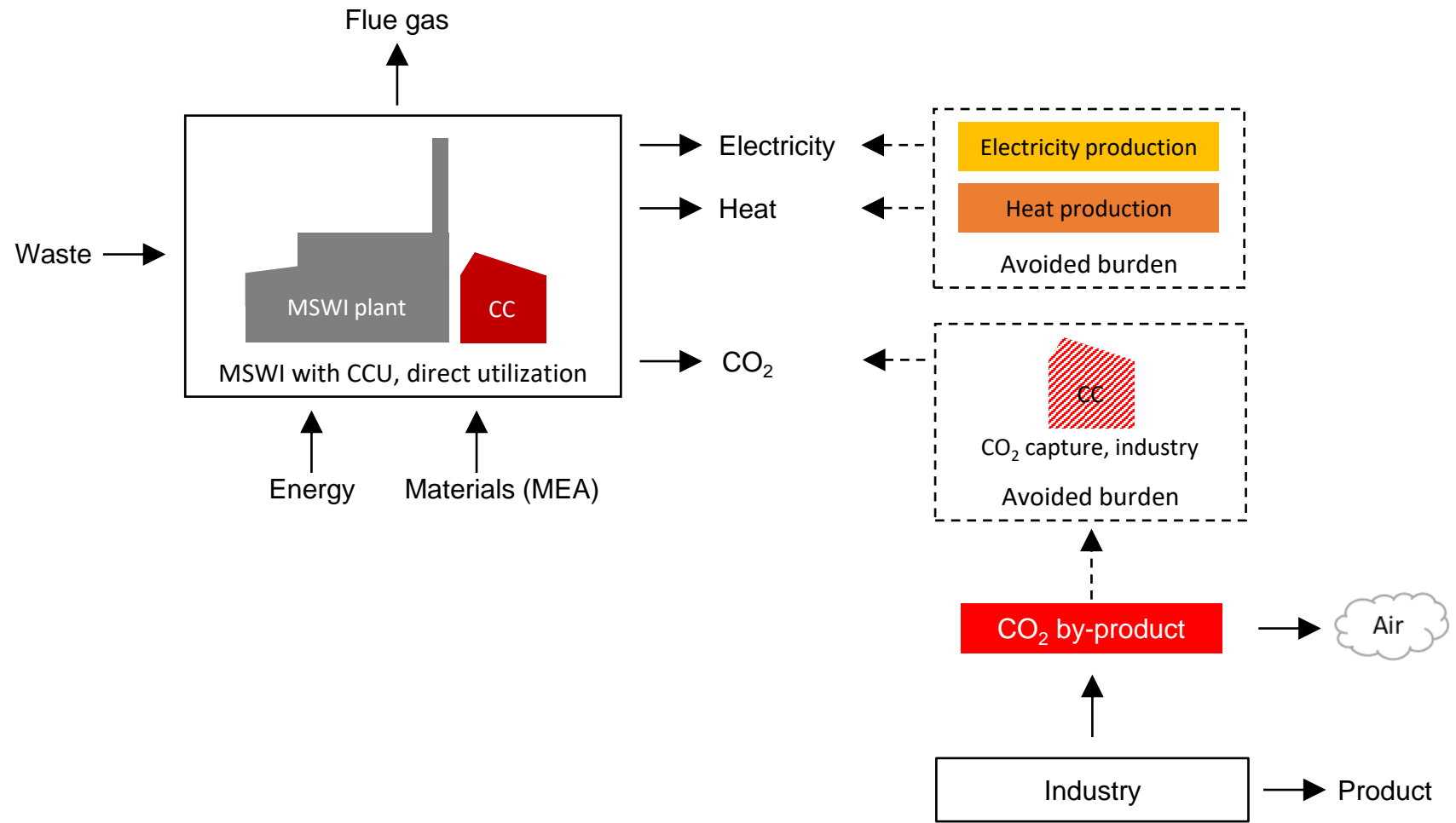


# Direct utilization: local use





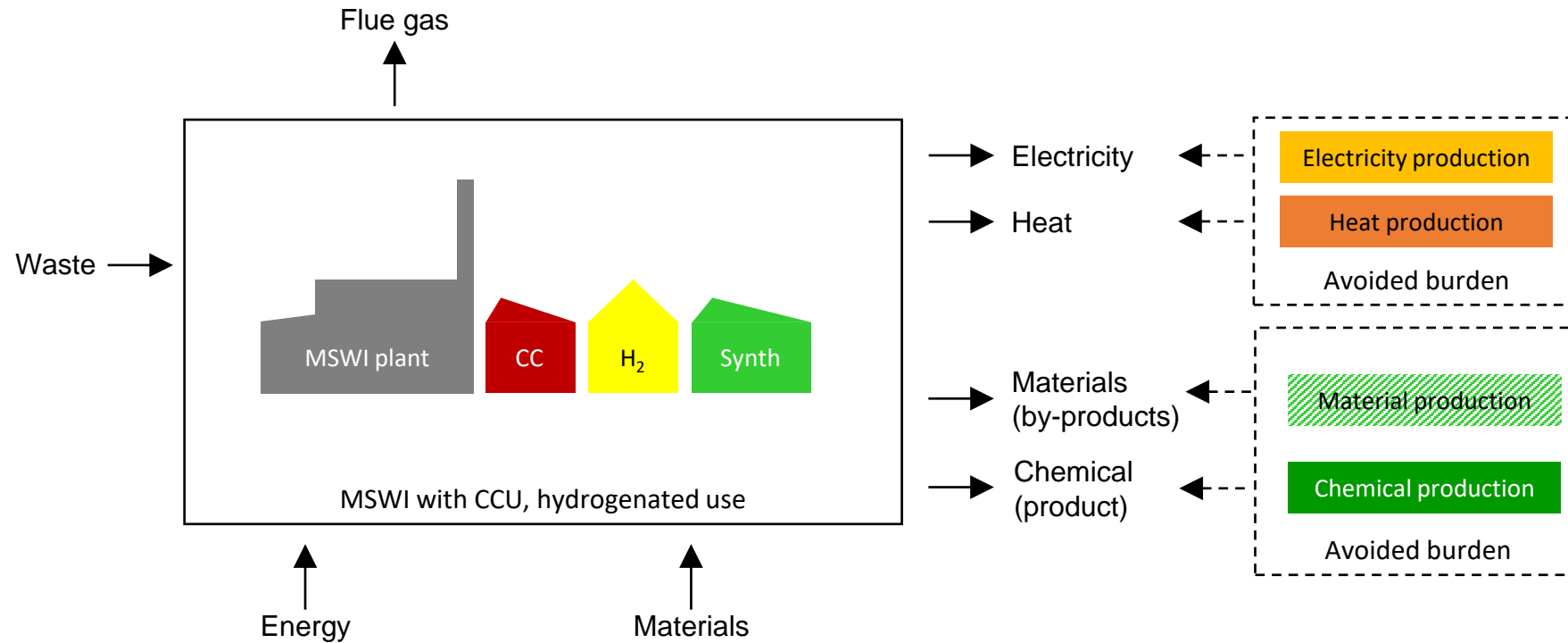
# Direct utilization: market use



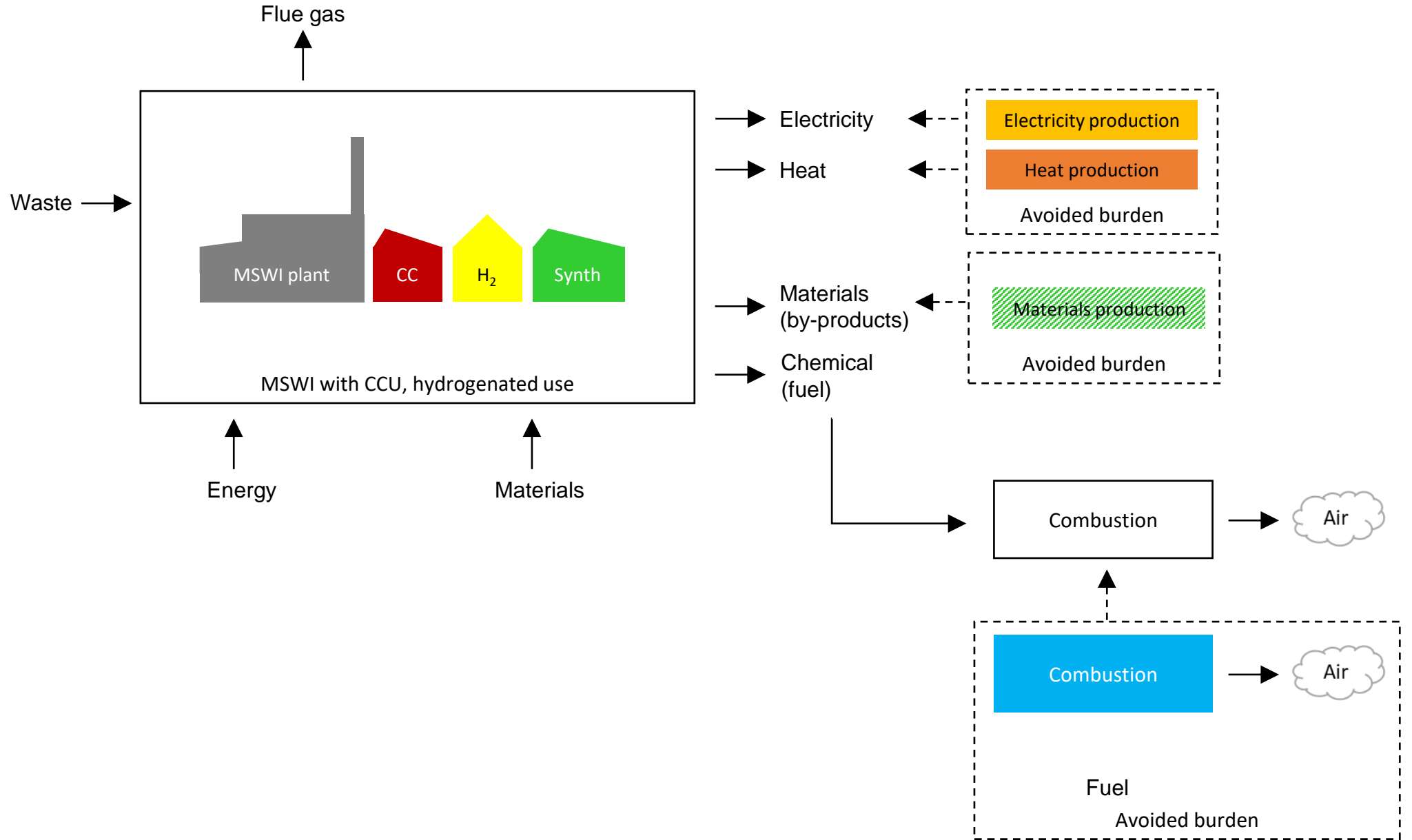
# Hydrogenated utilization

	CH <sub>4</sub>	CH <sub>3</sub> OH	DME	HCOOH
Names	Methane	Methanol	Dimethyl Ether	Formic Acid
	Methyl Hydride	Methylalcohol	Methoxymethane	Methanoic Acid
Chemical composition	CH <sub>4</sub>	CH <sub>3</sub> OH	CH <sub>3</sub> OCH <sub>3</sub>	HCOOH
		CH <sub>4</sub> O	C <sub>2</sub> H <sub>6</sub> O	CH <sub>2</sub> O <sub>2</sub>
H/C ratio in production process including H <sub>2</sub> O formation	8	6	6	4
Molar mass, g/mol	16.04	32.04	46.07	46.03
Heat of combustion, kJ/mol	890	726	1455	255
Heat of combustion, kJ/g	55.5	22.7	31.6	5.5
Boiling point, °C	-164	65	-24	101
Melting point, °C	-182	-98	-141	8
Density (liquid), kg/l	0.42	0.73	1.97	1.22
Liquefaction	Liquid below -150°C at pressures above 46 bar	Liquid at ambient temperature and pressure	Liquid at ambient temperature at pressures above 6 bar	Liquid at ambient temperature and pressure
Storage and transport	Like natural gas	Like alcohol	Like propane gas	Like weak acid
Used as baseline chemical	Yes	Yes	Yes	Yes
Used as fuel	Yes,	Yes	Yes	Yes,
	in combustion			in fuel cells
Other uses		Anti-freeze		Leather industry
				Dyeing
				Rubber industry
				Food and fodder
Climate change characterisation factor if released into the atmosphere, kg CO <sub>2</sub> -eq/ kg product	28	None	None	None
Carbon footprint of common production method, kg CO <sub>2</sub> -eq/ kg product	0.5	1.1	2.1	4.3

# Hydrogenated utilization: chemicals



# Hydrogenated utilization: fuels



# Production of chemicals

1000 kg product	Methane	Methanol	DME	Formic acid
<b>Inputs</b>				
CO <sub>2</sub> (kg)	3000	1450	2000	1000
H <sub>2</sub> (kg)	500	190	260	0
H <sub>2</sub> O (kg)	0	0	0	1410
Electricity (kWh)	60 <sup>a</sup>	20 <sup>a</sup>	30 <sup>a</sup>	12000
Steam/heat (GJ)	0	0	0	26
<b>Outputs</b>				
Product (kg)	1000	1000	1000	1000
H <sub>2</sub> O (kg) -reusable	2500	570	1175	180
H <sub>2</sub> (kg) – byproduct	0	0	0	92
O <sub>2</sub> (kg) - byproduct	0	0	0	1090
CO <sub>2</sub> (kg) - loss	250	75	85	40
Heat/steam (GJ)	9.0 <sup>b</sup>	1.4 <sup>c</sup>	2.5 <sup>d</sup>	0



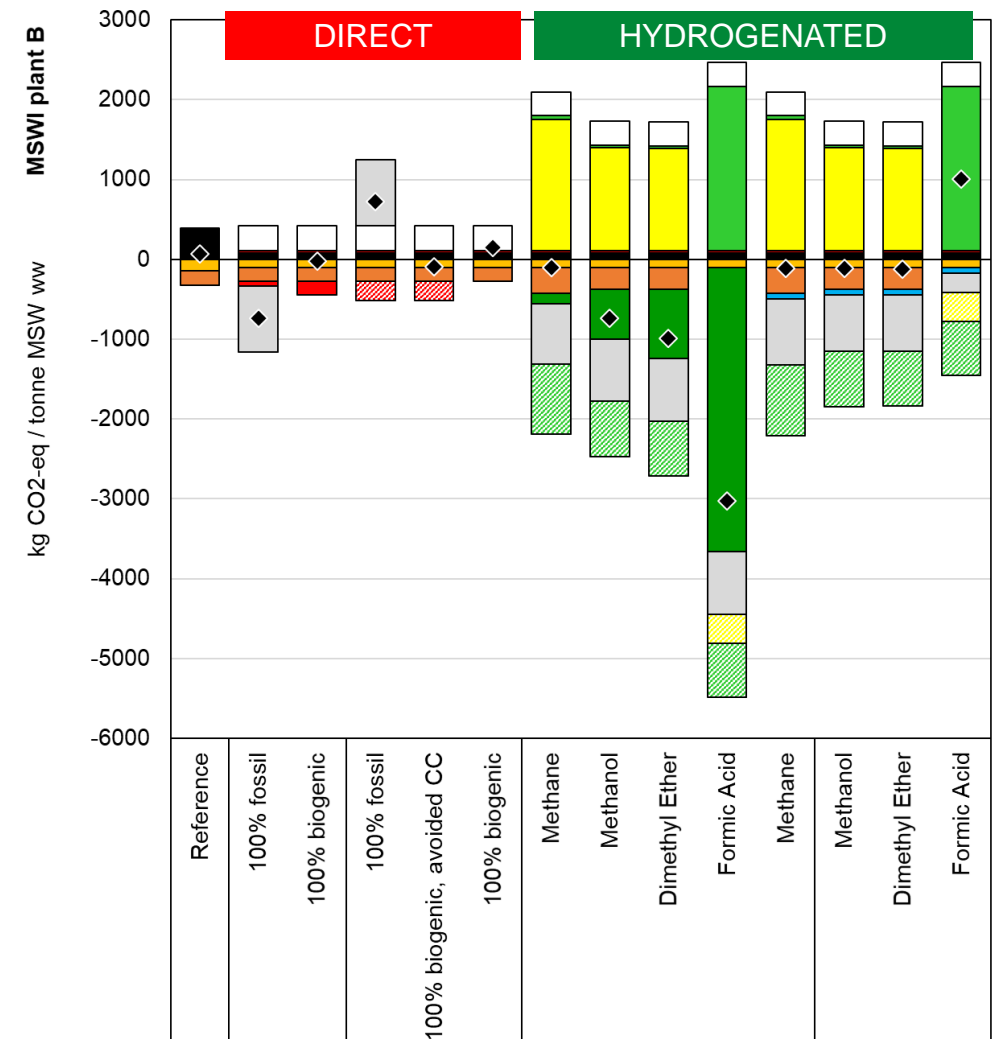
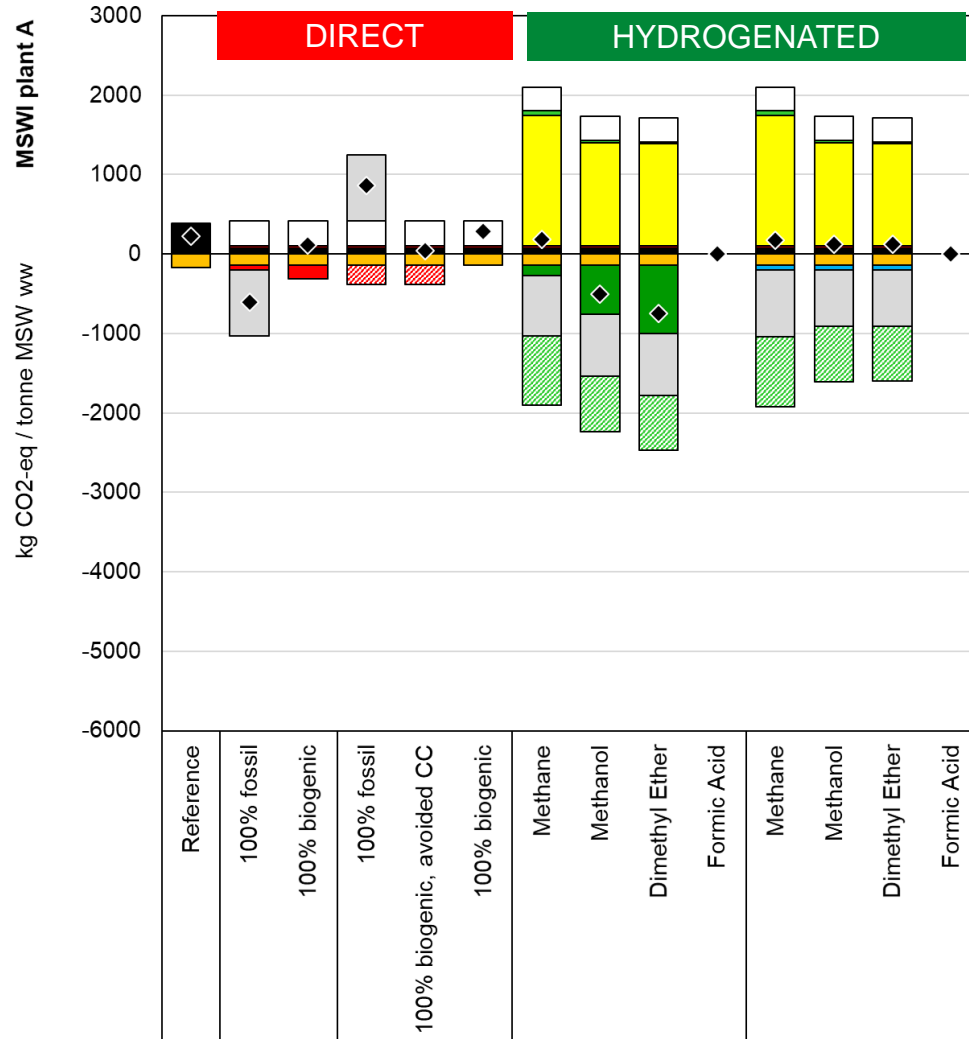
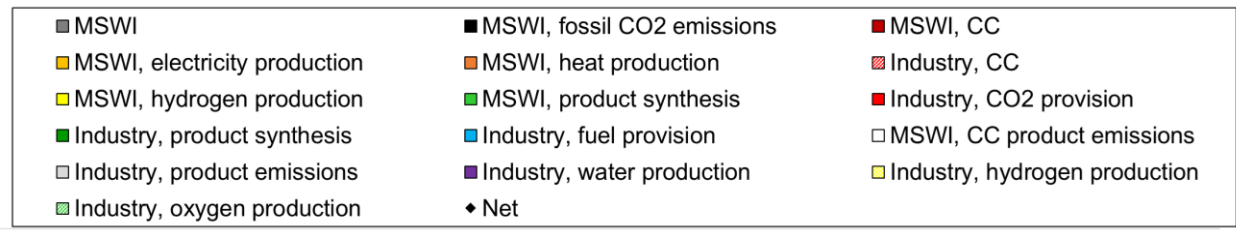
# Production of hydrogen

Flows for production of 1000 kg of H <sub>2</sub>	Hydrogen production (AEC electrolysis)
<b>Inputs</b>	
H <sub>2</sub> O k(g)	16800
Electricity (kWh)	58000
<b>Outputs</b>	
H <sub>2</sub> (kg)	1000
H <sub>2</sub> O (kg)	7200
O <sub>2</sub> (kg)	8400
Heat/steam (GJ)	62.4

# Technical outcome per ton of waste

MSWIB	Unit	Ref	Methane	Methanol	DME	Formic Acid
MSWI, net electricity	kWh/tonne ww	618	-7650	-5945	-5897	-9553
MSWI, net heat	GJ/tonne ww	8	14	12	12	0
MSWI, CO <sub>2</sub> air emissions, fossil	kg CO <sub>2</sub> /tonne ww	373	82	72	69	69
MSWI, CO <sub>2</sub> air emissions, biogenic	kg CO <sub>2</sub> /tonne ww	595	131	115	111	109
CC, ancillary material use (MEA)	kg MEA/tonne ww	0	4	4	4	4
H <sub>2</sub> production, water use (ultrapure)	kg H <sub>2</sub> O/tonne ww	0	-2303	-1811	-1796	0
H <sub>2</sub> production, oxygen by-product	kg O <sub>2</sub> /tonne ww	0	1152	905	898	897
H <sub>2</sub> production and synthesis, water by-product	kg H <sub>2</sub> O/tonne ww	0	1673	1099	1253	148
Synthesis, hydrogen by-product	kg H <sub>2</sub> /tonne ww	0	0	0	0	76
Synthesis, product	kg product/tonne ww	0	274	567	411	823
Synthesis, CO <sub>2</sub> loss, fossil	kg CO <sub>2</sub> /tonne ww	0	26	16	13	13
Synthesis, CO <sub>2</sub> loss, biogenic	kg CO <sub>2</sub> /tonne ww	0	42	26	21	20
Industry, product	kg product/tonne ww	0	-274	-567	-411	-823
Industry, oxygen	kg O <sub>2</sub> /tonne ww	0	-1152	-905	-898	-897
Industry, hydrogen	kg H <sub>2</sub> /tonne ww	0	0	0	0	-76
Industry, water	kg H <sub>2</sub> O/tonne ww	0	-1673	-1099	-1253	-148

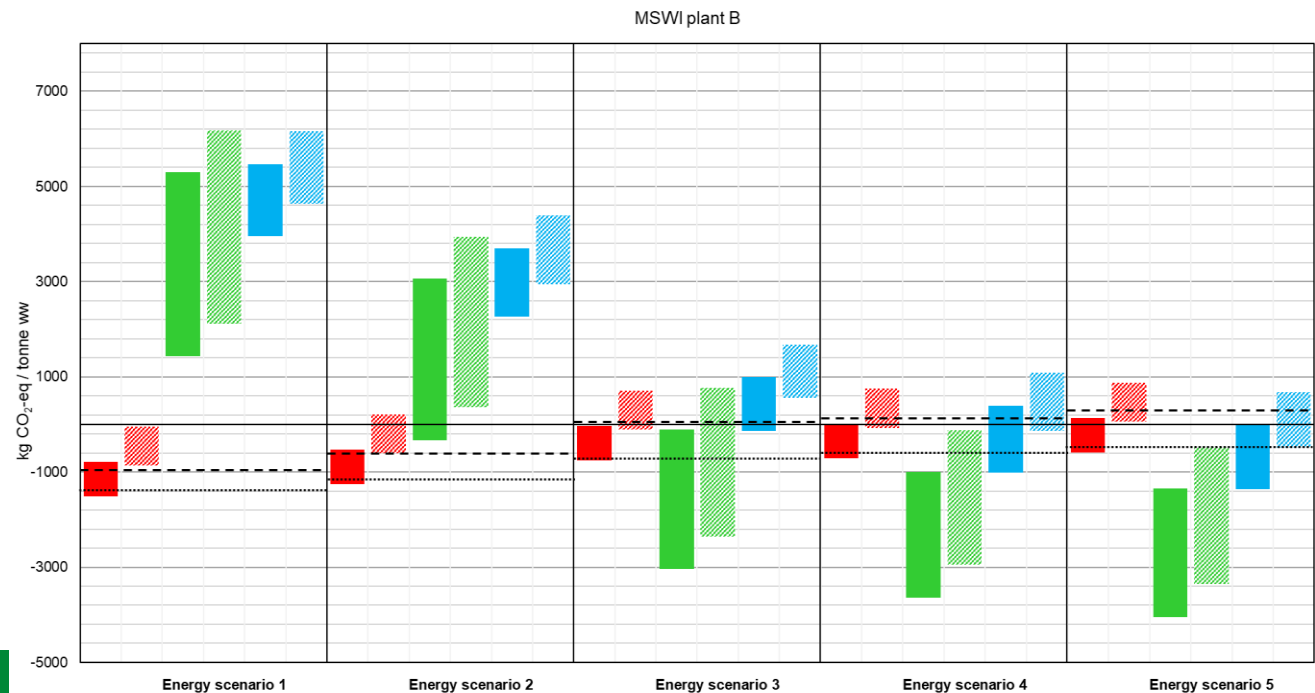
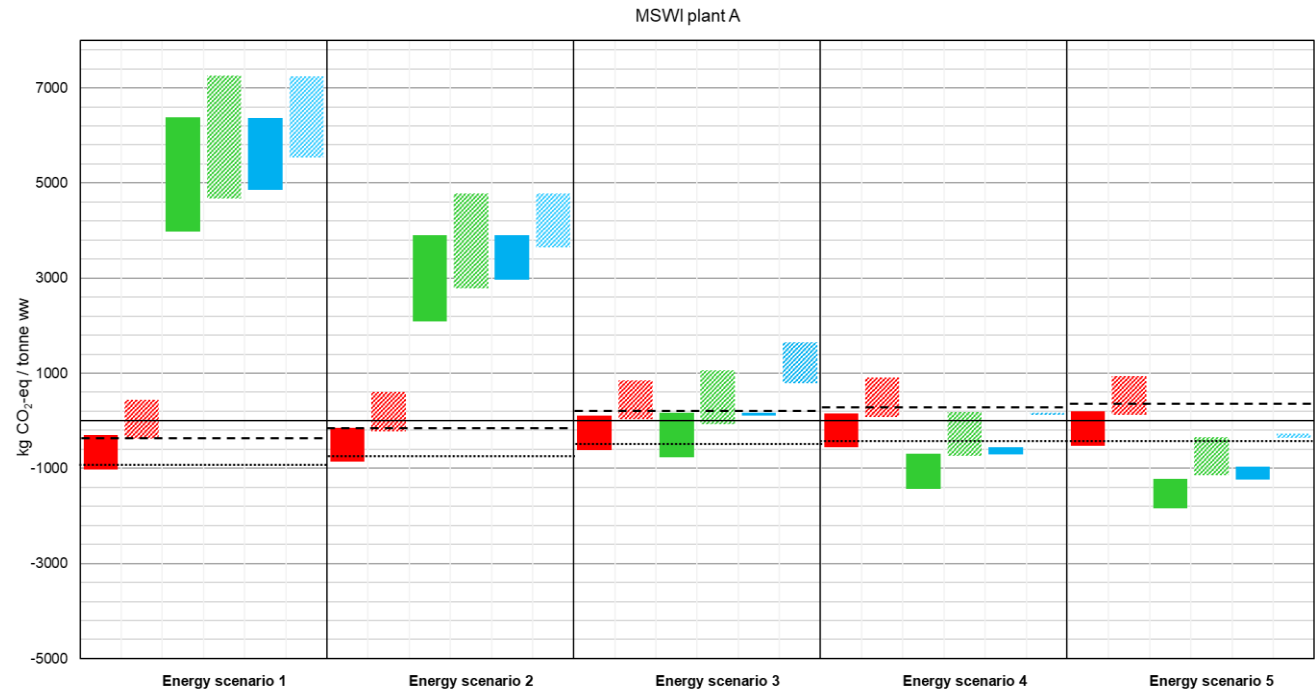
# Climate change impacts



# Climate change impacts

## Varying energy scenarios

**With 5000 kWh/tonne MSW of electricity pulled from the grid for CCU, an incinerator treating 500,000 tonnes of waste per year import about 2500 GWh electricity per year. In Denmark, a good land-based wind turbine delivers about 10 GWh electricity per year, corresponding to the annual consumption of about 2500 family houses. This suggests that the CCU at a large incinerator may need electricity from **250 wind turbines**, corresponding to the electricity used by 600,000 family houses. If we assume an average of three persons per family home, and 270 kg per person of waste incinerated annually, the MSWI plant will use the same amount of electricity as the amount of electricity used domestically by the number of people it services with waste management.**



- Direct utilization, local
- ▨ Direct utilization, market
- Hydrogenated utilization, chemicals, with market for oxygen byproducts
- ▨ Hydrogenated utilization, chemicals, without market for oxygen by-product
- Hydrogenated utilization, fuels, with market for oxygen byproducts
- ▨ Hydrogenated utilization, fuels, without market for oxygen by-product
- Reference MSWI
- MSWI with CC and storage



## Conclusion/ Discussion

Our results suggest:

- From a climate perspective it does make sense to CC at MSWI: Early phases as CCS evt as CCU if direct local uses can be identified, later when power is fully wind turbine based production of chemicals looks attractive
- Methanol and DME looks most promising

However do our assumtrions stay relevant in the long term?:

- What are the substitution wrt chemicals in the non-fossil future?
- Maybe the long term reference will be alternative fuels and even further in the future what is the cheapest cost of CC in any industry

MSWI image wise:

- Plastic becomes less impotent for the MSWI image because impacts will be savings
- But the less fossil content, the larger the saving



## CCS

Bisinella, V., Hulgaard, T., Riber, C., Damgaard, A., Christensen, T.H. (2021)  
**Environmental assessment of carbon capture and storage (CCS) as post-treatment technology in waste incineration.**  
*Waste Management*, 128, 99-113.

## CCU

Christensen, T.H. and Bisinella, V. (2021)  
**Climate change impacts of introducing carbon capture and utilization (CCU) in waste incineration.**  
*Waste Management*, 126, 754-770.

## CCS applied

Bisinella, V., Nedenskov, J., Riber, C., Hulgaard, T., Christensen, T.H. (2021)  
**Environmental assessment of amending the Amager Bakke incineration plant in Copenhagen with carbon capture and storage (CCS).**  
*Waste Management & Research*, .40, 79-95

