

BASF We create chemistry

H₂ – a key element today and in the future for BASF

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BASF – We create chemistry

Our chemistry is used in almost all industries

- We combine economic success, social responsibility and environmental protection
- Sales 2022: €87.3 billion
- EBIT before special items 2022: €6.9 billion
- Employees (as of December 31, 2022): 111,481
- 6 Verbund sites and 239 other production sites
- Around 82,000 customers from various sectors in almost every country in the world



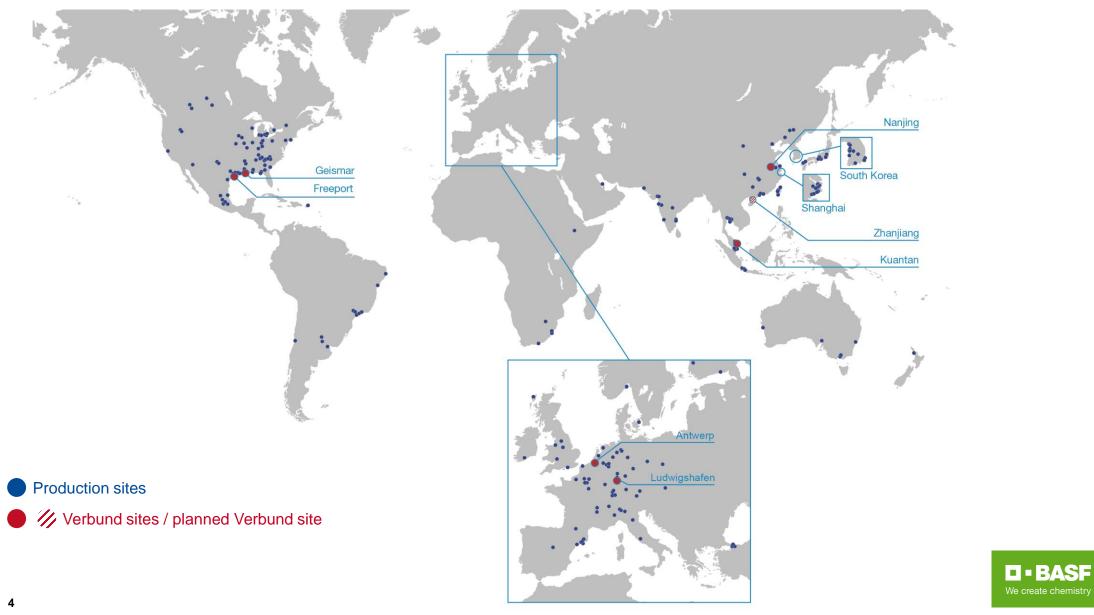


BASF's segments





BASF worldwide: Production and Verbund sites



We create chemistry for a sustainable future – BASF's emission targets

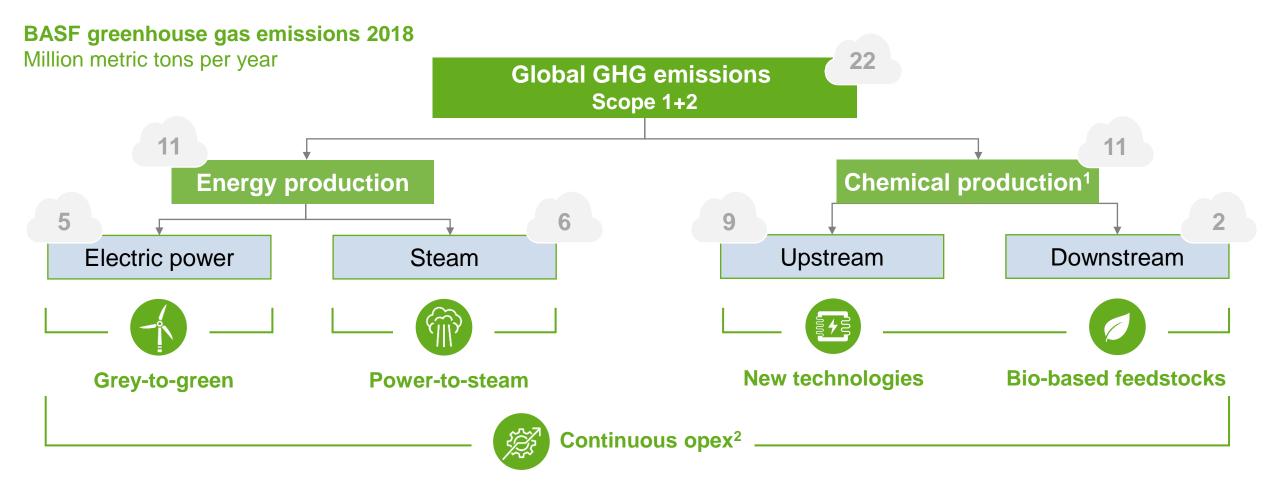
2030

25% CO₂ emissions reduction (compared with 2018)*

2050 net zero CO2 emissions*



No downstream transformation without upstream transformation







Hydrogen as a raw material

- Hydrogen is an irreplaceable raw material for many products in the chemical industry, but the current production process is associated with high CO₂ emissions.
- BASF pursues a make-and-buy approach to secure access to low-carbon hydrogen
- BASF is scaling up new technologies for production of clean hydrogen
- Availability and price of hydrogen as critical success factors

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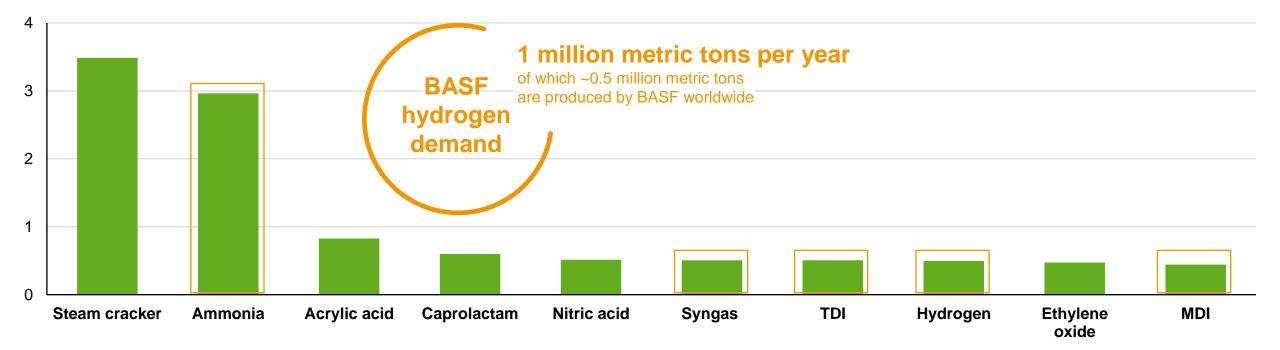
A clear legal framework is key, which allows technology-open transformation

The use of hydrogen as a raw material is a key lever for CO₂ emission reduction across several technologies



Greenhouse gas emission profile of BASF technologies

Energy and chemistry emissions, million metric tons per year*

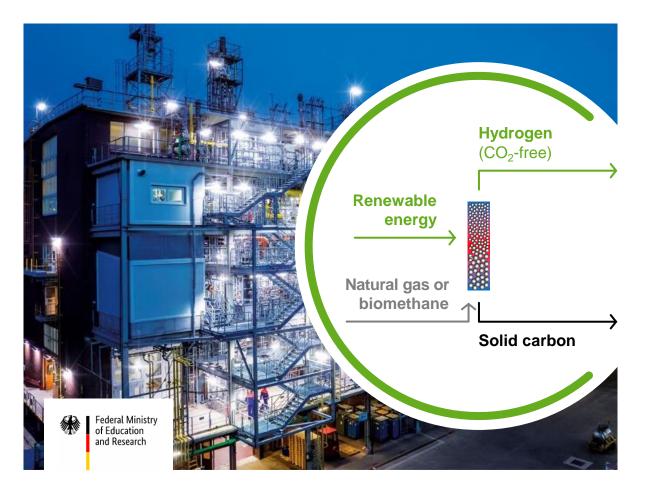


Achieving the production of CO_2 -free hydrogen will tackle 2 to 3 million metric tons of our CO_2 emissions across several technologies



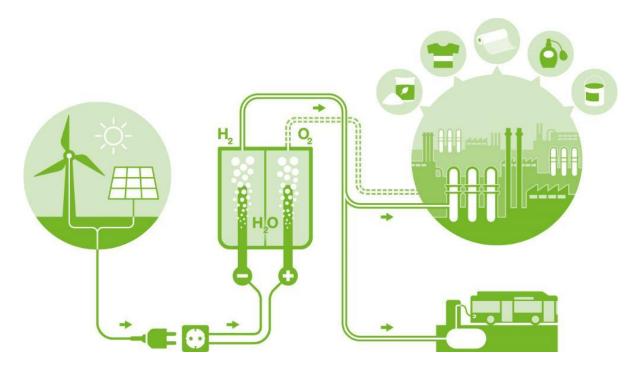
Methane pyrolysis^{*} – process innovation to reduce CO₂ emissions





- Methane pyrolysis is a low-emission technology. In this innovative process, (bio)methane is split directly into hydrogen and solid carbon
- Test plant at the Ludwigshafen site in trial operation
- Key challenges are process technology and control
- Methane pyrolysis requires around 80% less electricity than water electrolysis and is virtually carbon-free if renewable energy is used

Water electrolysis in Ludwigshafen – BASF's Hy4Chem project





- In water electrolysis, water is split directly into its two components, hydrogen and oxygen
- If the required energy comes from renewable sources, the process is carbon-free
- We are working with Siemens Energy on a project for the construction of a PEM (proton exchange membrane) water electrolyzer with a capacity of 54 megawatts
- Hydrogen to be used in BASF Verbund and for local community hydrogen mobility market
- BASF applied for funding by the German Federal Ministry for Economic Affairs and Climate Action (BMWK)



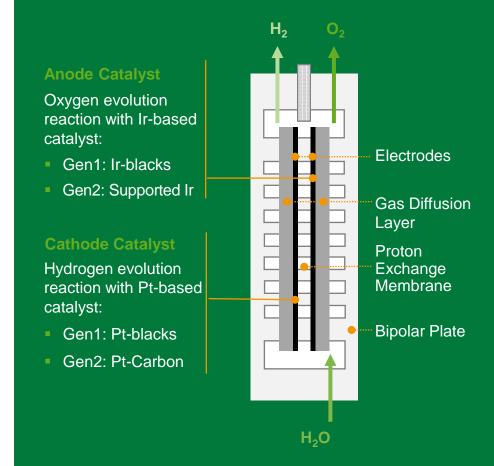
PEM-Electrolysis needs high-performance, low PGM*-loading catalysts

- Ir- and Pt-based electrocatalyst are key to high efficiency and long-term stability of the electrolyzer stack
- Today's limited Ir-supply and the projected demand growth for PEM-electrolyzers call for the development of catalysts that combine lower Ir-loadings with higher efficiency

BASF catalysts offer various benefits to our customers:

- ✓ Low-PGM loading
- High efficiency & performance
- High corrosion stability for enhanced lifetime
- ✓ PGM handling: sourcing, supply and recycling

PEM Electrolyzer and catalysts





Increasing importance of Hydrogen as an energy vector

- After renewable power, H₂ is the energy vector of choice for the transformation
- Future BASF hydrogen demand as energy vector is expected to be significant
- Build-up of gas infrastructure must be accelerated to meet huge demand.
- A pipeline connection to BASF sites in Europe is key to access cost-competitive H₂ and BASF is looking at different supply routes to ensure the connection

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